


# Health-based recommended occupational exposure limit for Pyridine

Dutch expert committee on occupational  
standards (met Nederlandstalige samenvatting)

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Health-based recommended occupational exposure limit for Pyridine

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# Health-based recommended occupational exposure limit for Pyridine

Dutch expert committee on occupational standards (met Nederlandstalige samenvatting)

Centraal Instituut voor Arbodienst en Arbeidsbescherming

1993-1-1

Health-based recommended occupational exposure limit for Pyridine. The expert committee on occupational standards (Dutch expert committee for occupational standards) has determined a health-based recommended occupational exposure limit for Pyridine of 0.1 mg/m<sup>3</sup> (8-hour time-weighted average). This limit is based on the results of a toxicological study in which the lowest observed adverse effect level (LOAEL) was found to be 0.1 mg/m<sup>3</sup> (8-hour time-weighted average). The expert committee has determined that this LOAEL is the basis for the recommended occupational exposure limit.

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**ANNEX 1**

**DECOS and SCG Basis for an Occupational health Standard Pyridine.**  
**Margareta Warholm, ARBETE OCH HÄLSA 1991:49**

Margareta Warholm, Arbete och Hälsa 1991:49. Detta är en översättning av den svenska utgåvan av denna rapport. Den svenska utgåvan är tillgänglig på Arbetsförhållanden, Box 101, S-100 01 Stockholm, Sverige. Detta är en översättning av den svenska utgåvan av denna rapport. Den svenska utgåvan är tillgänglig på Arbetsförhållanden, Box 101, S-100 01 Stockholm, Sverige.

## Pyridine

### 1. Samenvatting fysische en chemische gegevens

CAS nr: 110 - 86 - 1

Pyridine is een zeer brandgevaarlijke kleurloze tot lichtgele vloeistof met een zeer onaangename geur. Het is mengbaar met water, alcohol, ether en andere organische oplosmiddelen. Het is een goed oplosmiddel voor organische en anorganische verbindingen. Pyridine is lichter dan water en de damp is zwaarder dan lucht. Het reageert heftig met oxidatiemiddelen. Pyridine heeft een relatieve molecuulmassa van 79,1, een kookpunt van 115,3°C, een smeltpunt van -41,6°C, een relatieve dichtheid van 0,98 (water = 1), een dampspanning van 2,66 kPa (25°C) en een relatieve dampdichtheid van 2,73 (lucht = 1).

Conversiefactoren (25°C): 1 ppm = 3,23 mg/m<sup>3</sup>; 1 mg/m<sup>3</sup> = 0,309 ppm

### 2. Monitoring

Voor de omgevingsmonitoring op de werkplek wordt verwezen naar de methode van het NIOSH (1977). Bij deze methode wordt gebruik gemaakt van persoonlijke monsternamen apparatuur en van een gaschromatograaf. Er is geen methode bekend voor biologische monitoring van werknemers die blootgesteld zijn aan pyridine.

### 3. Grenswaarden

De huidige MAC-waarde voor pyridine in Nederland is 15 mg/m<sup>3</sup> (5 ppm), TGG-8 uur. Deze grenswaarde geldt ook in Duitsland, het Verenigd Koninkrijk, Zweden, Frankrijk en de Verenigde Staten. Zweden adviseert ook een kort-durende grenswaarde van 35 mg/m<sup>3</sup> (10 ppm).

### 4. Toxicokinetiek

Opname kan plaatsvinden via het maagdarmkanaal, longen en huid. Na opname wordt de stof verspreid naar verschillende weefsels/organen. De hoogste concentraties na de opname worden gevonden in bloed, nieren en urine.

Er zijn weinig gegevens over het metabolisme van pyridine bij de mens. Orale toediening van een heel lage dosis pyridine bij twee proefpersonen resulteerde in het voorkomen van een aantal metabolieten in de urine. Ongeveer 30 en 10% van de toegediende dosis wordt terug gevonden in de vorm van respectievelijk pyridine N-oxide en N-methylpyridinium ion in 24-uurs urinemonsters. Uit experimenten met proefdieren blijkt dat de biotransformatie van pyridine species- en dosis-afhankelijk is. Pyridine wordt uitgescheiden via longen, huid, faeces en urine in de vorm van vrije-basen en metabolieten.

### 5. Effecten

Effecten bij proefdieren: Bij kortdurende blootstelling heeft pyridine een narcotische werking, tevens worden symptomen van irritatie van de slijmvliezen waargenomen. Pyridine heeft een lage acute toxiciteit

en de orale LD50 ligt hoger dan 1 g/kg bij kleine proefdieren. Effecten op de nier en lever worden gevonden na langdurende blootstelling via de voeding. Een dosis van 50 mg/kg/dag toegediend via maagsonde gedurende 90 dagen bij ratten veroorzaakt verlaging van het lichaamsgewicht en bij 70% van de dieren wordt ontsteking van de lever gevonden. Muizen die die dosis gedurende 3 maanden met drinkwater (2,0 µg/ml) kregen toegediend, vertonen toename van de vet peroxidatie in de striatum en het cerebellum deel van de hersenen. Na continue inhalatoire blootstelling van ratten aan een concentratie van 1 mg/m<sup>3</sup> (0,3 ppm) gedurende twee maanden werden lichte effecten gevonden op de functie van het centrale zenuwstelsel en de samenstelling van bloedewitten. Een samenvatting van de dosis-effect relatie bij proefdieren is beschreven in Tabel 2.

Effecten bij de mens: Een samenvatting van de dosis-effect relatie voor de mens blootgesteld aan pyridine is beschreven in Tabel 1. Volgens de OSHA ligt de dodelijke luchtconcentratie van pyridinedamp ongeveer bij 3600 ppm. Blootstelling aan gemiddeld 125 ppm in lucht, 4 uur per dag gedurende één tot twee weken, veroorzaakte lichte effecten op het centrale zenuwstelsel en het maagdarmkanaal. Zeven gevallen van chronische pyridine vergiftiging werden vermeld in een chemisch bedrijf met lucht concentraties van 20 tot 42 mg/m<sup>3</sup> pyridine. De symptomen zijn hoofdpijn, tijdelijke duizeligheid, overspannenheid, slapeloosheid en maagdarmklachten zoals misselijkheid en braken.

## 6. Evaluatie en advies

Geconcludeerd kan worden dat de kritische effecten bij kortdurende blootstelling, de prikkelende werking op de slijmvliezen van de bovenste luchtwegen en ogen zijn, en de bedwelmende werking op het centrale zenuwstelsel. Anderzijds zijn de nier en lever de kritische organen na langdurende blootstelling. Uit gegevens bij de mens blijkt dat een walgingwekkende geur voorkomt bij een concentratie van 32,3 mg/m<sup>3</sup> (10 ppm) pyridine in lucht. De reukgrens ligt tussen 0,04 en 2,6 mg/m<sup>3</sup> (0,013 - 0,82 ppm). Effecten op het centrale zenuwstelsel treden op bij concentraties tussen 19,4 en 42 mg/m<sup>3</sup> (6 - 13 ppm). Dit betekent dat een kortdurende advieswaarde beneden 19,4 mg/m<sup>3</sup> zou moeten liggen.

Voor de langdurende blootstelling aan pyridine zijn twee dierexperimentele gegevens van belang. Bij een blootstelling aan 10 of 50 ppm (32,3 of 161 mg/m<sup>3</sup>) pyridine damp in lucht bij ratten, 5 dagen/week, gedurende 6 maanden wordt toename van het gewicht van de lever gevonden. In een andere experiment, waarbij ratten via maagsonde doseringen van 0,25, 1, 10, 25 en 50 mg/kg/dag pyridine gedurende 90 dagen kregen, werd toename van het levergewicht gevonden bij de doseringen van 10 en 25 mg/kg/d. Hieruit kan worden geconcludeerd dat de NOAEL op 1 mg/kg/d zou kunnen liggen. Extrapolatie naar de mens levert een concentratie van 7 mg/m<sup>3</sup> (2 ppm) pyridine damp in lucht (aangenomen dat: gemiddeld lichaamsgewicht = 70 kg, ademvolume gedurende een werkdag van 8 uur = 10 m<sup>3</sup> en 100% resorptie in maagdarmkanaal en longen). Ter compensatie van de onzekerheden en interspecies variaties stelt de WGD voor een veiligheidsfactor 10 te gebruiken. De gegevens over de carcinogeniciteit van pyridine zijn ontoereikend.

Aan de hand van de boven vermelde gegevens stelt de WGD een gezondheidskundige advieswaarde voor van 1 mg/m<sup>3</sup> (0,3 ppm), TGG - 8 uur en een "H" notatie.

Datum van afsluiting: juni 1993



## 1. INTRODUCTION

This document is a co-production of the Dutch Expert Committee on Occupational Standards (DECOS) of the Dutch Directorate-General of Labour and the Swedish Criteria Group for Occupational Standards (SCG) at the Swedish National Institute of Occupational Health. It is the product of an agreement (signed in 1988) between both groups to write joint scientific criteria documents which could be used by the national regulatory authorities both in the Netherlands and in Sweden.

The draft as prepared by dr. Margareta Warholm was first reviewed by the SCG and thereafter by the DECOS. The resulting document, as published by the Swedish National Institute of Occupational Health (Arbete och Hälsa 1991:49), is included in the present document as Annex 1.

In order to be able to serve as a basis for setting a Dutch MAC-value some extra chapters are supplemented to the Swedish version by the DECOS, resulting in the underlying document. Most important is the addition of a "Recommended Health Based Occupational Exposure Limit".

## 2. ANALYTICAL METHODS

### 2.1 ENVIRONMENTAL MONITORING

Methodology according to NIOSH (1977):

Instrumentations: Personal monitoring equipments and gas chromatography instrumentation.

Principle: A known volume of air is drawn through a charcoal tube to trap the organic vapors present. The charcoal in the tube is then transferred to a small stoppered sample container, and the analyte is desorbed with methylene chloride. An aliquot of the desorbed sample is injected into gas chromatograph. The area of the resulting peak is determined and compared with areas obtained for standards. This method is validated over the range of 7.59-30.4 mg/m<sup>3</sup> at an atmosphere temperature and pressure of 22 degrees Celsius and 763 mm Hg, using a 100 l sample. Under the conditions of sample size (100 l) the probable useful range of this method is 1.5 - 45 mg/m<sup>3</sup> at a detector sensitivity that gives nearly full deflection on the stripchart recorder for a 4,5 mg sample. The method is capable of measuring much smaller amounts if the desorption efficiency is adequate. Desorption efficiency must be determined over the range used.

### 2.2 BIOLOGICAL MONITORING

No methods are applicable at present.

Ref: NIOSH. NIOSH manual of analytical methods. Second edition. VOL 3 (1977 PP S161-1-S161-8).

3. GUIDELINES AND STANDARDS

COUNTRY (organization)	CONCENTRATION		COMMENTS
	mg/m <sup>3</sup>	(ppm)	
<u>The Netherlands</u> 1989	15	(5)	twa - 8 hour
<u>Sweden</u> 1987	16	(5)	twa - 8 hour
	35	(10)	short term
<u>United States</u>			
ACGIH 1990 - 1991	16	(5)	twa - 8 hour
OSHA-PEL	15	(5)	twa - 8 hour
NIOSH-REL	15	(5)	twa - 8 hour
<u>Germany</u> 1991	15	(5)	twa - 8 hour
<u>United Kingdom</u>			
HSE 1991		(5)	twa - 8 hour
		(10)	short-term
<u>France</u> 1986		(5)	twa - 8 hour
		(10)	short - term

- Ref:
- Arbeidsinspectie: De nationale MAC-lijst 1992, P145
  - National Swedish Board of Occupational Safety and Health. Occupational Exposure Limit Values (1987)
  - American Conference of Governmental Industrial Hygienists. Threshold Limit Values for Chemical Physical Agents and Biological Exposure Indices (1990-1991)
  - ACGIH. Guide to Occupational exposure values - 1990
  - Deutsche Forschungsgemeinschaft. Maximale Arbeitsplatzkonzentrationen und Biologisch Arbeitsstofftoleranzwerte 1991
  - Health Safety Executive. Occupational Exposure Limits 1991
  - IHRS. Values limites pour les concentrations des substances dangereuses, 1986.

#### 4. PREVIOUS EVALUATION BY (INTER)NATIONAL BODIES

The ACGIH concluded that the TLV of 5 ppm, as a TWA, and the STEL of 10 ppm should be low enough to prevent systemic effects from exposure to pyridine, provided skin absorption is not permitted, but may not eliminate complaints about the odour.

This conclusion seems to be in contrary with the data reported beforehand in their document. It was considered that the most important effect of pyridine after exposure by inhalation was chronic poisoning, the critical organs being the liver, kidney and bone marrow. Mild symptoms may be found after exposure to 10 ppm pyridine. It was also reported that long-term poisoning with mild symptoms of the central nervous system occurred in workers of a plant where pyridine concentrations ranged from 6 to 12 ppm. This means that there is almost no safety margin between the TLV and the occurrence of symptoms.

The DFG does not have any documentation on the MAK of pyridine.

Ref: ACGH. Documentation of the threshold limit values and biological exposure indices. Fifth edition. Cincinnati, Ohio, 1986.

## 5. EVALUATION OF HUMAN HEALTH RISK

### 5.1 Groups at extra risk

No data are available to determine specific groups at extra risk.

### 5.2 Assessment of health risk

In human health risk assessment due to occupational exposure to pyridine vapour, two types of exposure associated effects are of relevance. It may be summarized that the critical effects after short-term exposure is the irritation of the mucous membranes of the upper respiratory tract and eyes, and acute effects on the central nervous system. On the other hand the critical organs after long-term exposure seem to be the kidney and the liver.

Human data indicate that objectionable odour is experienced at a level of  $32.3 \text{ mg/m}^3$  (10 ppm) pyridine. The range of odour threshold lies between  $0.04$  and  $2.6 \text{ mg/m}^3$  (0.013 - 0.82 ppm) pyridine, according to Ellenhorn and Barceloux (1988). Effects on the central nervous system with symptoms of headache, nervousness, sleeplessness and occasional digestive tract problems are reported at levels of  $19.4$  to  $42 \text{ mg/m}^3$  (6 - 13 ppm) pyridine in workroom air. From these events it may be concluded that the short-term occupational exposure limit should be below  $19.4 \text{ mg/m}^3$ .

Long-term exposure studies on experimental animals and human data indicate that kidney and liver are the critical organs. The human data could not be used for risk assessment because the doses used are out of proportion and the patients (epileptics) used them as drug for anticonvulsant treatment.

There are two animal experiments important for risk assessment. In one experiment with rats exposed by inhalation to levels of either 10 or 50 ppm ( $32.3$  or  $161.5 \text{ mg/m}^3$ ) pyridine vapour, 5d/w, 6 months, increased liver weights were found (Anonymous, 1986; Gehring 1983).

In the other study (Anderson, 1987) pyridine was administered to rats by gavage at doses of 0.25, 1.0, 10, 25 and  $50 \text{ mg/kg}$  for 90 days. At the highest dose inflammatory hepatic lesions were found as well as increased absolute and relative liver weight.

Ref: Ellenhorn, MJ and Barceloux DG Medical toxicology, diagnosis and treatment of human poisoning (1988) Elsevier, New York.

Increased relative liver weights were found in female, but not in male rats, receiving 10 or 25 mg/kg pyridine. From this study it may be concluded that the NOAEL for these effects is 1 mg/kg b.w. per day. Extrapolation from rat to man means that in man (standard: 70 kg) breathing 10 m<sup>3</sup> volume of air in 8 hour the NOAEL should be 7 mg/m<sup>3</sup> (2 ppm). In this estimation the extrapolation is based on the assumption that 100% of pyridine is absorbed through the digestive tract as well as respiratory tract since the compound is completely soluble in water and penetrates very well through the skin.

In this kinetic model some assumptions have been made, e.g. the use of gavage as a method of administration, which is not the normal method of intake in real life, and the 100% absorption through the respiratory tract. The DECOS endeavors to compensate these uncertainties by using a safety factor. The data on the carcinogenic potential of pyridine are too meagre to be of use in the risk assessment. The DECOS proposes to use a safety factor of 10 and an upward roundoff to arrive at a health-based recommended occupational exposure limit of 1 mg/m<sup>3</sup> (0.3 ppm) t.w.a. 8 - hour pyridine vapour with a skin notation.

### 5.3 Health-based Recommended Occupational Exposure Limit

1 mg/m<sup>3</sup> (0.3 ppm) pyridine, time weighted average 8 hours, with a skin-notation.

## 6. RECOMMENDATION FOR RESEARCH

1. Epidemiological studies are required on groups of workers with long-term exposure to pyridine vapour.
2. Studies on the kinetic of pyridine in humans.
3. Development of biological monitoring methods.

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VETENSKAPLIG SKRIFTSERIE

1991:49

**DEC and SCG Basis for an Occupational  
Health Standard**

**PYRIDINE**

*Margareta Warholm*

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## DEC and SCG Basis for an Occupational Health Standard

# PYRIDINE

Margareta Warholm

*The Swedish National Institute of Occupational Health employs over 300 scientists in research on the work environment. The research is led by 30 professors. The Institute does mostly applied research, but some questions also require basic research.*

*The scientific competence of the Institute is concentrated in six areas: Physiology, Chemistry, Medicine, Psychology, Technology and Toxicology. This wide competence provides solid support for the Institute's cross-disciplinary approach.*

*The Institute is responsible for training safety engineers, physical therapists and psychologists, as well as doctors and nurses for the industrial health services.*

*Another of the Institute's responsibilities is disseminating information on occupational health research.*

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## PREFACE

An agreement has been signed by the Dutch Expert Committee for Occupational Standards (DEC) of the Dutch Directorate-General of Labour and the Swedish Criteria Group at the Swedish National Institute of Occupational Health (SCG). The purpose of the agreement is to write joint scientific criteria documents for occupational exposure limits. These limits will be developed separately by the two countries according to their different national policies.

This document on health effects of pyridine is a product of the agreement. The document was written by Dr. M. Warholm from the National Institute of Occupational Health in Solna, Sweden, and was reviewed by the Dutch Expert Committee as well as by the Swedish Criteria Group.

J. J. Kolk  
Chairman  
Dutch Expert Committee

B. Holmberg  
Chairman  
Swedish Criteria Group



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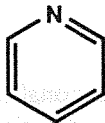
## Physical and chemical properties

CAS nr: 110-86-1

Synonyms: azabenzene, azine

Molecular formula:  $C_5H_5N$

Structural formula:



Molecular weight: 79.1

Melting point:  $-41.6\text{ }^{\circ}\text{C}$

Boiling point:  $115.3\text{ }^{\circ}\text{C}$

Relative density: 0.98

Vapor pressure: 2.66 kPa at  $25\text{ }^{\circ}\text{C}$

Vapor density: 2.73

Flash point:  $20\text{ }^{\circ}\text{C}$

Solubility: completely soluble in water, alcohol, ether, acetone and benzene.

pKa: 5.19

pH value: 8.5 (0.2 M solution in water)

Conversion factors:  $1\text{ ppm}=3.23\text{ mg/m}^3$ ;  $1\text{ mg/m}^3=0.309\text{ ppm}$  at  $25\text{ }^{\circ}\text{C}$

Pyridine is a highly flammable, colourless (when pure) or slightly yellow liquid with a characteristic disagreeable odor and a burning taste. It is miscible with water, alcohol, ether, oils, and many other organic compounds, and is slightly alkaline in reaction. It is a good solvent for both inorganic and organic compounds. Pyridine is lighter than water and its vapors are heavier than air. Pyridine forms an azeotropic mixture with three moles of water (59 vol %-41 vol % water), which boils at  $92\text{-}93\text{ }^{\circ}\text{C}$ . Contact of pyridine with strong oxidizers may cause fires and explosions. Contact with strong acids will cause violent splattering. Toxic gases and vapors (such as oxides of nitrogen and carbon monoxide) may be released in a fire involving pyridine.

In humans, the air odor threshold for pyridine in most individuals is less than 0.2 ppm (3, 4, 14), but olfactory fatigue occurs quickly (6). Pyridine becomes objectionable to unacclimatized individuals at 10 ppm (60). The irritation threshold for pyridine is 700 ppm (4, 65). This value is defined as the lowest concentration that could be distinguished from pure air by a general anosmic, i.e. by a person who has no olfactory nerve sensation, but whose trigeminal nerve sensitivity is intact. The pungency thresholds for pyridine in three anosmics described in (18) are of similar magnitude.

## **Production and uses**

Pyridine is produced synthetically or by fractional distillation of coal tar residues.

Pyridine is used as a solvent in e.g. pharmaceutical and polycarbonate resin industries. It is particularly useful as a solvent in processes where HCl is evolved (it forms stable salts with strong acids). A main use of pyridine is in the production of agricultural chemicals, such as the herbicides paraquat and diquat, and the insecticide chlorpyrifos. Other uses include the production of piperidine, the manufacture of pharmaceuticals, and in dyeing of textiles. Minor quantities are used for the denaturation of alcohol and antifreeze mixtures, and as a flavoring agent (33, 59, 60).

## **Occurrence**

Pyridine is found in tobacco smoke and roasted coffee, as well as in the leaves and roots of *Atropa belladonna*. It is also found in coal tar, wood oil and as a component of creosote oil (59).

In the U.S.A. in 1978, air concentrations of pyridine in work places where pyridine was manufactured or used as an intermediate ranged from 0.008 to 1 ppm. Monitoring of technicians working with pyridine in quality control and research and development laboratories of one of the pyridine manufacturers showed that the highest TWA concentration measured over a 6-hour period was 0.09 ppm (61).

## **Toxicokinetic data**

### *Absorption*

Pyridine is absorbed from the gastrointestinal tract, intraperitoneal cavity, and lungs. It is also rapidly absorbed through intact skin (60). Based on its aqueous and lipid solubility, the absorption of pyridine is expected to be rapid (61). In rats, uptake by tissues following i.p. administration increased with dosage (72).

### *Distribution*

Pyridine gains access into all tissues, but the largest concentrations are found in the blood, kidney and urine (66). Based on the acute toxic effects of pyridine it can be assumed that pyridine is distributed to the central nervous system, liver and kidneys (61).

### Biotransformation

Pyridine has been shown to be oxidized to several metabolites *in vivo* and *in vitro*. These include pyridine N-oxide, 3-hydroxypyridine, 2-pyridone, and 4-pyridone; an N-methylpyridinium ion metabolite is also formed (64).

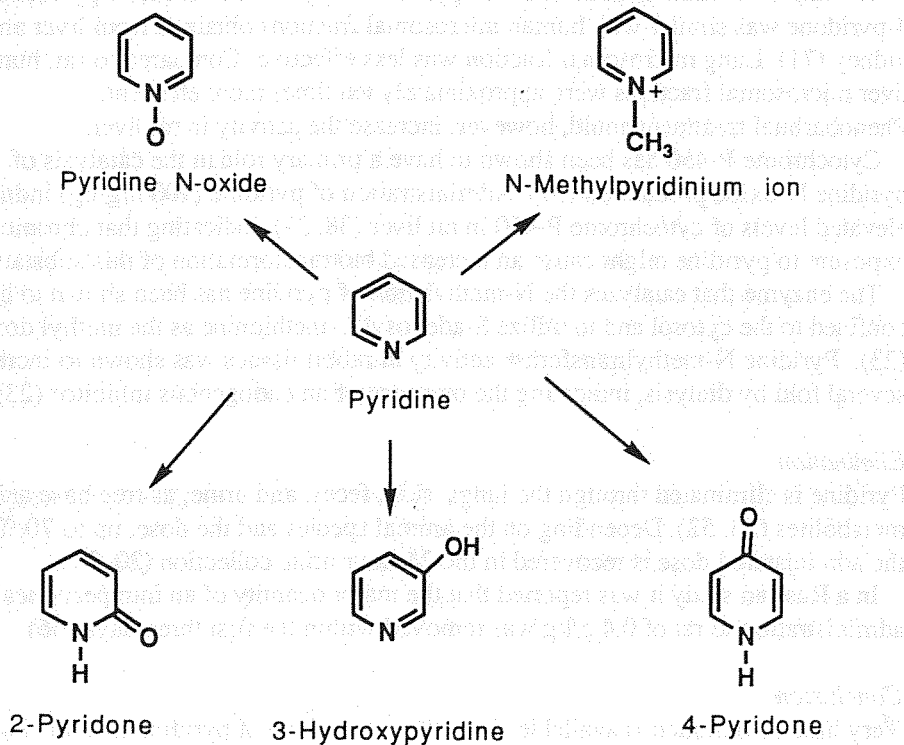


Figure 1. Structure of pyridine and some of its mammalian urinary metabolites.

Damani and co-workers investigated urinary metabolites of  $^{14}\text{C}$ -pyridine in eight different species, including man (21). The total amount of  $^{14}\text{C}$  excreted in the 24-h urine samples ranged from about 50% to 80% of the administered dose, which was 7 mg/kg i.p., except for the two male volunteers, who received approximately 50  $\mu\text{g}/\text{kg}$  p.o. A distinct species difference in the metabolism of  $^{14}\text{C}$ -pyridine was noted. For example, in cat and rabbit 15% to 25% of the administered dose was recovered as unchanged pyridine, whereas only minor amounts of unmetabolized substance was found in hamster, gerbil, rat, and mouse. 4-Pyridone seems to be a major metabolite in rabbit and guinea pig, whereas the most common metabolite in hamster is pyridine N-oxide. In some species a large portion of urinary metabolites is unidentified. In man, N-oxidation and N-methylation appeared to be important routes of metabolism.

In rat and guinea pig, the urinary excretion of pyridine is clearly dose-dependent (20). Thus, in the rat, the percentage of 24-h urinary metabolites dropped from 70% at an administered dose of 1 mg/kg to only 6% at a dose of 500 mg/kg. There was also a decline in the proportion of the dose excreted as N-methylpyridinium. In contrast, N-oxidation assumes greater importance as a route of metabolism at higher doses of pyridine (21, 22).

The rate of oxidative metabolism of pyridine to pyridine N-oxide, 2-pyridone or 4-pyridone was similar with human microsomal fractions obtained from liver and kidney (71). Lung microsomal fraction was less effective. Compared to rat, human liver microsomal fractions were approximately ten times more efficient. Phenobarbital treatment could, however, increase the activity in rat liver.

Cytochrome P-450 has been shown to have a primary role in the catalysis of pyridine N-oxide production (51). Administration of pyridine (100 mg/kg) induced elevated levels of cytochrome P-450 in rat liver (38, 39), indicating that chronic exposure to pyridine might cause an increased biotransformation of this substance.

The enzyme that catalyzes the N-methylation of pyridine has been shown to be confined to the cytosol and to utilize S-adenosyl-L-methionine as the methyl donor (23). Pyridine N-methyltransferase activity in rabbit tissues was shown to increase several fold by dialysis, indicating the presence of an endogenous inhibitor (23).

#### *Elimination*

Pyridine is eliminated through the lungs, skin, feces, and urine, as free base and as metabolites (33, 52). Depending on the animal species and the dose, up to 70 % of the administered dose is recovered in the 24-hour urine collection (20, 21).

In a Russian study it was reported that the major quantity of an intraperitoneal administration to rat of 0.4 g/kg was removed within the first three days (66).

#### *Conclusion*

Very little information is available about the metabolism of pyridine in man. Two males received a very low dose of pyridine p.o. and only some urinary metabolites were analyzed. Approximately 30 % and 10 % of the administered dose was recovered as pyridine N-oxide and N-methylpyridinium ion, respectively, in the 24-h urine samples (21). Studies using experimental animals have shown that the biotransformation of pyridine is both species- and dose-dependent. In addition, there is no information about other than urinary metabolites. Jori *et al.* (33) concluded in 1983 that the knowledge of the kinetics and metabolism of pyridine is totally inadequate and that information on metabolic fate is based either on very old publications or on recent studies which were not specifically designed to study pyridine metabolism. No *in vivo* studies on man have appeared thereafter.

#### **Toxicological mechanisms**

No studies that were aimed at investigating the mechanisms of pyridine toxicity have been found in the literature. It is possible though that pyridine interferes with endogenous pyridine nucleotide metabolism. Pyridine may also act on nicotinic receptors (43). It has been suggested that the metabolism of pyridine increase its toxicity. The N-oxide of pyridine is a more potent toxicant than the parent compound (61), and the metabolite 3-hydroxypyridine is a potential teratogen (51).

Clugnet & Laget (17) suggest that the effects of pyridine on the motor behaviour of experimental animals may be due to acetylation in the blood of pyridine to 3-acetylpyridine, which affects cerebellar functions. However, that such an acetylation occurs has not been established.

## General toxicological effects in animals

### *Short-term exposure*

Pyridine has a narcotic action when administered to experimental animals. Effective doses by any route produce weakness, ataxia, salivation, and unconsciousness, but convulsions are uncommon. With exposure to pyridine vapors there are symptoms of moderate mucous membrane irritation. Pyridine has relatively low acute toxicity. In most cases the oral LD<sub>50</sub> for small animals exceeds 1 g/kg (60). There is a general depression of the CNS and death results from respiratory paralysis (43, 59). In various bird species LD<sub>50</sub> for pyridine has been estimated to exceed 1 g/kg (62).

The acute dermal lethal dose of pyridine for rabbits was 1120 mg/kg (7). This value is only slightly higher than the oral LD<sub>50</sub> (for rat), indicating that pyridine is effectively absorbed through the skin

For rats, inhalation of pyridine vapor for 4 h yielded an LD<sub>LO</sub> of 4000 ppm (49). Concentrated vapor (23200 ppm) caused 100% mortality within 1.5 h (60).

### *Long-term exposure*

Rats receiving 0.1% pyridine in the diet (approximately 50 mg/kg body weight/day, (59)) gradually lost weight and died within 2-4 weeks (12). The livers and kidneys revealed acute lesions and some of the livers exhibited well marked cirrhosis. Addition of 0.5% methionine to the diet decreased the toxicity of pyridine. Another group of rats in this study received 0.164% in the diet of the pyridine metabolite methylpyridinium ion (as chloride), this dosage being equivalent to 0.1% pyridine. All six animals survived and showed continual growth. Two were sacrificed on the 32nd day and were found to have histologically essentially normal livers and kidneys.

In another study it was shown that pyridine toxicity decreased when the choline and casein content of the diet increased (11). Pyridine administered daily by subcutaneous injection or by stomach tube proved to be less toxic than when given in the diet (11).

Sprague-Dawley rats (10 male and 10 female rats in each group) were given pyridine by gavage at dosages of 0, 0.25, 1.0, 10.0, 25.0, or 50.0 mg/kg for 90 days (5). The male rats given the highest dose of pyridine (50 mg/kg) showed statistically significant decreased body weights (days 57-85) and total weight gain. Inflammatory hepatic lesions were found in 70% of the animals compared to 10% in the controls. In females, 50 mg/kg caused mild hepatotoxicity with significantly increased absolute and relative liver weights. Inflammatory hepatic lesions appeared in 20% of the females in this group. Increased relative liver weights were also found in females, but not in males, given 10 or 25 mg/kg. In this study no significant histopathological or clinical findings related to the neurological systems were found that could be definitely attributed to exposure to pyridine.

The MTD for rats injected subcutaneously, twice weekly for four weeks, was found to be less than 180 mg/kg. Twice weekly injections of 100 mg/kg for one year did not produce any changes in mortality relative to controls (45).

Four rabbits dosed orally for 5-7 days with 44 mg/kg pyridine showed no noticeable effects (58).

Administration to mice of pyridine in the drinking water (2.0 µl/ml) for 3 months induced a significant increase in lipid peroxidation in striatum and cerebellum (57).

Repeated exposure of rats (7 h/day, 5 days/week for 6 months) to either 10 or 50 ppm pyridine vapor did not change either the growth rate or mortality when compared to controls; however increased liver weight/body weight ratios were noted in the treated rats (8, 28).

Continuous inhalational exposure of rats for two months at 1 mg/m<sup>3</sup> (0.3 ppm) caused minor effects on CNS function and blood protein composition, while 0.1 mg/m<sup>3</sup> was without effect (41).

Chronic exposure of guinea pigs to vapors of commercial-grade pyridine for 4 months (3 h/day) at a concentration of 1000 mg/m<sup>3</sup> caused loss of body weight, depression of body temperature and hypochromic anemia (66). A histopathological examination of some of the animals showed hepatic cirrhosis, focal necrosis and fatty liver cells.

Intragastric administration to albino rats of pyridine at doses of 0.25 and 0.125 mg/kg during 100 days led to decreases in body weights and disturbances in the functional state of the liver. A pyridine dose of 0.0125 mg/kg did not affect the body weight of the rats. The speed with which conditioned reflex was produced in mice was not altered by exposure to 0.0125 mg/kg (66).

## General toxicological effects in humans

Browning states that pyridine is less toxic in animals than in human beings (13).

Most of the effects observed in man are transient, involving the CNS and gastrointestinal tract. Despite its relatively large industrial and some medical use, reports of injurious effects of pyridine in man have been relatively uncommon (15).

### *Short-term exposure*

A fatal case from accidental ingestion of "half a cupful" of commercial pyridine was reported by Helme in 1893 (31). Initially there was vomiting. Later there was slight cyanosis, a choking sensation, precordial and abdominal pain, and a raised temperature, pulse and respiration. There was no abnormality of the urine. Delirium and acute congestion of the lungs preceded death which occurred 43 hours after the ingestion. Autopsy revealed that the major lesions were inflammatory and were situated in the respiratory tract, the oesophagus and the stomach. There were a few fatty patches in the liver.

Browning notes a case of acute narcosis in a man who had been employed in cleaning a tank-waggon that had contained pyridine (no further details were given) (13).

Another case of acute poisoning is noted in (60). In this case a woman had been decontaminating a pyridine spillage for 15 to 20 minutes. Symptoms did not occur for ten hours and then intensified until the third day. They included speech disorders and what was recorded as "rather diffused cortical affliction" which

receded after thiamine therapy. There were no symptoms from the upper respiratory tract.

The probable lethal dose is 0.5-5.0 g/kg (54), and a  $LD_{50}$  for man was 500 mg/kg p.o. (48).

OSHA states that a vapor concentration in air of 3600 ppm is immediately dangerous to life and health (54).

#### *Long-term exposure*

Ludwig reported in 1935 two cases of pyridine poisoning from its industrial use (44). In one case a chemist, who had worked with crude pyridine for six months, suffered from disturbance of equilibrium, facial paralysis and attacks of loss of consciousness. These symptoms disappeared when the man stopped working with pyridine. In the other case, the symptoms after two years' exposure were paralysis of the ocular muscles with nystagmus, facial paralysis, hemiparesis with anaesthesia to heat and paraesthesia of the left side of the face, right-sided excessive perspiration, cerebellar ataxia, bladder paralysis, difficulty of hearing and neuralgic headache. The author considered these symptoms similar to Wernicke's pseudo-encephalitis.

In the treatment of epilepsy, short-term administration of 0.6 ml pyridine 3 or 4 times a day was found by Pollock *et al.* to cause minor complaints of anorexia, nausea, occasional vomiting and headache, faintness, weakness and mental depression (58). Treatment was continued in two cases for up to two months, resulting in severe liver and kidney damage (in one case fatal).

Transient symptoms from the CNS and the GI tract, without associated liver and kidney damage, have occurred in individuals exposed to pyridine concentrations averaging 125 ppm, 4 hours a day, for 1 to 2 weeks (60).

Teisinger (67) reported seven cases of chronic pyridine poisoning in a chemical plant where air samples showed concentrations of 20-42 mg/m<sup>3</sup>. The symptoms were headache, temporary vertigo, nervousness, sleeplessness, together with occasional digestive troubles, particularly nausea and vomiting. No objective clinical signs were found in any of the cases.

In a Russian review it was noted that an examination of 88 production employees inhaling pyridine vapors at concentrations of 10 to 20 mg/m<sup>3</sup> showed signs of myocardial degeneration, symptoms of chronic gastritis, functional CNS changes, and sensitivity disturbances of the polyneuritic type. The incidence of these alterations increased with the duration of employment (66).

### **Effects on organs**

#### *Skin and mucous membranes*

With exposure to pyridine vapors there are symptoms of moderate mucous membrane irritation (60). Liquid pyridine is severely irritant to skin (66). Skin irritation in rabbits is described as mild after application of 10 mg/24 h (open Draize test) or 500 mg/24 h (standard Draize test) (50).

In rabbits, the primary irritation index was estimated to 4.8 (maximum possible score=8) (26). Necrosis was observed on the scarified skin of all animals. On the intact skin individual differences were important; very slight to severe erythema with rare necrosis, no or slight oedema. In all cases no regeneration occurred after



72 h. Histopathological examination showed both lesions of necrosis and regeneration phenomena.

Pyridine acts as a photosensitizer in skin (27). Skin eruptions provoked by light, and limited to exposed surfaces, have been induced by pyridine in humans (9).

In the "maximization test" by Kligman, performed on humans, the allergenic potential of pyridine was classified as weak (grade 1) (25).

### *Eye*

Pyridine vapor is irritating to the eyes and lids. Pure pyridine applied to rabbit eyes caused moderate injury (graded 7 on a scale of 1 to 10 after 24 h). However, an aqueous solution (0.08 M) of pyridine applied to the denuded cornea or injected into the corneal stroma caused no reaction (29).

Dutertre-Catalla *et al.* (26) classify pyridine as severely irritating to the eye. Pyridine is not only irritating but also corrosive and necrotizing for the eye.

In a human experiment, 0.098-0.11 mg/m<sup>3</sup> (0.0295-0.033 ppm) pyridine temporarily affected the sensitivity of the eye to light (15, 41). The practical significance of this finding is unclear.

### *Lung*

Rats were exposed to pyridine vapor at a concentration of 15 mg/m<sup>3</sup> for 60 days. During the first six days this exposure resulted in a depression of lipid formation in pneumocytes, with a reduction of protein synthetic activity of cells and a reduction in the amount of phospholipids in the surfactant. Signs of adaptation appeared the following weeks, but after the 36th day there was a fall of phospholipids in the alveolar cells and in the surface-active lining of the alveoli (56).

No data from human exposure is available.

### *Digestive organs*

Rats exposed to pyridine (0.2 mg/l) and Ca<sup>2+</sup> (100 mg/l) in their drinking water showed a thickening of the mucous membrane folds of the stomach, abundant mucous in the stomach with small amounts in the duodenum, as well as catarrhal symptoms and ulceration. When Mg<sup>2+</sup> was added instead of Ca<sup>2+</sup> no catarrhal symptoms or ulcerations were seen (40).

In humans, chronic exposure to pyridine may affect the GI tract and cause digestive troubles (44, 60, 67).

### *Liver and kidney*

Chronic exposure to pyridine is toxic to the liver and kidneys. The kidney is less sensitive to pyridine-induced damage than the liver is (28).

In studies by Baxter *et al.*, already referred to (10, 11, 12), 0.1% pyridine in the diet given to rats caused death in 2-4 weeks. The principal lesions were found in the livers and the kidneys, and it was presumed that death was due to disturbances in the functions of these organs.

Rats given 50 mg/kg pyridine by gavage for 90 days showed inflammatory hepatic lesions in some of the animals (5). Increased liver weights were also noted.

Exposure to pyridine by inhalation may also cause liver damage. Guinea pigs exposed to 1000 mg/m<sup>3</sup> pyridine (3h/day, 4 months) exhibited toxic effects in the liver (66).

A single exposure of rats to 5000 or 10000 mg/m<sup>3</sup> for 40 minutes affected the kidneys' capacity to produce ammonia and their contribution to sustaining acid-alkaline balance in the body (66).

In a study where 17 anaesthetized dogs were given pyridine at various dose levels between 88 and 880 mg/kg body weight intravenously (LD<sub>5</sub> to LD<sub>50</sub>), significant and dose-dependent increases in serum GOT and blood urea were noted, as well as decreases in serum alkaline phosphatase, indicating liver and kidney damages (70).

In man, the treatment of epilepsy with pyridine (1.85 - 2.4 ml per day for up to two months) caused progressive parenchymal destruction of the liver in two cases (58).

### *Heart*

In the rat, intravenous administration of pyridine in doses between 1 and 15 mg/kg caused tachycardia and a small fall in blood pressure. Higher doses, up to 75 mg/kg, produced marked bradycardia and hypotension, with changes in the ECG (43). The vasodilator effect of pyridine is also referred to.

ACGIH notes that large doses of pyridine are toxic to the heart (2). It is not clear whether this refers to humans.

### *Nervous system*

Pyridine has a narcotic action when administered to experimental animals. Effective doses by any route produce weakness, ataxia, unconsciousness and salivation (60). Also in man cases with CNS symptoms, including headache, dizziness, and fatigue, have been reported from exposures to pyridine vapor (13, 44, 59, 67).

By intravenous administration the narcotic dose in rabbits is 110 mg/kg (13, 58).

In the Wistar rat, large, i.e. more than 100 mg/kg, intravenous injections of pyridine through the jugular vein caused alterations of the spontaneous neo- and archeo-cortical bioelectrical activities (17). These alterations were reversible. Chronic treatment of rats with intraperitoneal injections of more than 330 mg/kg pyridine affected the motor behaviour of the animals. The duration of this experiment is not specified.

In an experiment on dogs, intravenous administration of pyridine (88-880 mg/kg) affected the nervous system. Commonly observed manifestations were salivation, myosis, lacrimation, nasal secretion, micturition, cloudy cornea, apnoea and death by cardiac failure, which invariably followed administration of a dose of 880 mg/kg i.v. (70).

A Russian paper states that the minimal concentration of pyridine that changed the pattern of bending reflex in rabbits by a 4-minute exposure was 400 mg/m<sup>3</sup>. The lowest effective dose that acted on the unconditioned reflex response was 12.5 to 25 mg/kg (66).

Inhalation of high concentrations of pyridine (5000 or 10000 mg/m<sup>3</sup> for 40 minutes) caused accumulation of ammonia in the brain of the test rats 1 to 3 days after the exposure (66). As previously mentioned, pyridine in the drinking water (2.0 µg/ml for 3 months) caused increased lipid peroxidation in mouse brain (57).

In an *in vitro* system using cultured mouse neuroblastoma cells, pyridine inhibited the respiratory activity of the cells only at very high concentrations (ED<sub>50</sub>=40 mM) (69).

## Biochemical effects

Chronic administration of pyridine to rabbits and rats has been shown to induce cytochrome P-450, and to result in an increased biotransformation of pyridine and also of N-nitrosodimethylamine, alcohols and aniline (36, 51). A single injection of pyridine (10 to 200 mg/kg, i.p.) caused a dose-dependent induction of cytochrome P-450 (P450IIE1) in rat liver (38). Pyridine has also been shown to inhibit the metabolism, and thereby the clastogenicity, of benzene (30), as well as reversibly inhibit the respiratory burst in porcine and human neutrophils (32). Aromatic L-amino acid decarboxylase may be affected by pyridine (34), and pyridine has also been shown to change ammonia metabolism (66).

## Reproductive effects and effects of pregnancy

When injected into chicken eggs at a dosage of 20 mg per egg, pyridine caused typical muscular hypoplasia of the legs and, very rarely, skeletal abnormalities of the neck and face. These effects were decreased with nicotinamide supplementation (42).

Pregnant guinea pigs showed increased sensitivity to the toxic effects of pyridine at chronic exposure to 1000 mg/m<sup>3</sup> (66).

## Mutagenicity and genotoxicity

Santodonato (61) notes that pyridine was not mutagenic in an extensive assay using *Salmonella typhimurium* strains TA1535, TA1537, TA1538, TA1536, and TA100. The compound was tested at 1, 10, and 100 µg/plate with or without the addition of a metabolic activation system (from 7 rat tissues). EPA reports that assays in *Salmonella typhimurium* tester strains TA98, TA100, TA1535 and TA1537 were negative following pyridine incubation (max. 10 µl sample per plate) with or without added metabolic activation (47).

A significant increase in forward mutation frequency was seen in *S. typhimurium* strain TM677 with microsomal activation (35). The minimum dose with mutagenic effect was 6 mM. For comparison, benzo(a)pyrene was mutagenic at 4 µM. However, in another study, using the same strain and 25 mM pyridine as the highest concentration, no mutagenesis was reported (63).

In the *Drosophila* sex-linked recessive lethal test pyridine was inactive in a feeding test and gave questionable results after injection (68). Also in the L5178Y tk<sup>+</sup>/tk<sup>-</sup> mouse lymphoma cell forward mutation assay, pyridine was found not to be mutagenic (46).

Negative results were also obtained in a test on *Drosophila*, where pyridine was administered throughout the larval development of the flies at 0.01, 0.1 or 1 mM concentration in the food. Surviving treated adults, as well as the F<sub>1</sub> and F<sub>2</sub> progenies, were screened for tumors and developmental abnormalities (19).

In cultured Chinese hamster cells, Abe & Sasaki (1) reported a significant, but not dose-related, increase in the frequency of sister-chromatid exchanges, while no chromosome aberrations were found. The concentrations of pyridine used were 1, 2, and 5 mM.

In a spot test in *E. coli*, no increase in reverse mutants to Trp<sup>+</sup> were produced following pyridine exposure (55).

Pyridine caused a slight induction of mitotic aneuploidy in yeast at 0.8-1.3 % (73). It was suggested that pyridine interfered with tubulin assembly and disassembly causing a malfunctioning of spindle fiber microtubules.

Pyridine has also been tested for its ability to inhibit intercellular communication in the metabolic cooperation assay using Chinese hamster V79 cells and found to be negative (16).

A cooperative program on short-term assays for carcinogenicity in Japan 1973-1978 came to the conclusion that pyridine is not mutagenic (37). The assay systems used were *S. typhimurium* TA100 and TA98 (mutations), *Bacillus subtilis* (*rec* assay, without metabolic activation), hamster lung fibroblast cells (chromosome aberrations, sister chromatid exchanges), rat bone marrow cells (chromosome aberrations *in vivo*), and silk worms (mutations).

### Carcinogenicity

In a study by Mason et al. (45), pyridine was administered to rats (Fischer 344) by subcutaneous injection at 3, 10, 30, or 100 mg/kg twice weekly for one year. An additional six-month latency period was allowed prior to autopsy. The incidence of injection site, mammary or total tumors was not increased in pyridine-exposed animals compared to unexposed or vehicle controls.

Hepatic nodules were reported in some rats exposed to pyridine in the diet at 0.20-0.29 % for up to 4 months (11). Lesions were not invasive and no metastases were seen, indicating their benign nature.

Santodonato (61) notes that the National Toxicology Program has conducted a chronic carcinogenicity bioassay of pyridine in which the compound was administered by gavage to rats (F344) and mice (B6C3F1), and that this bioassay has been completed and the technical report been drafted (NCI/NTP Technical Report No. 283). The report has not yet been published though. It is probably this study which is referred to in ref. (24), where it is noted that pyridine caused a positive trend for mononuclear cell leukemia in male Fischer 344 rats in a 2-year study performed by the NCI/NTP. The incidences of leukemia were 23 %, 40 % and 45% in the control, low-dose (25 mg/kg) and high-dose (50 mg/kg) groups.

Pyridine treatment also accelerated the rate of tumor growth in a cellular transplant model for leukemia in Fischer rats (24). Pyridine was given in the drinking water at doses of 0, 0.5 or 1.0 mg/ml and these doses were meant to replicate those used in the 2-year study.

A long term animal study for carcinogenesis of pyridine was scheduled to start in 1990 (53).

### Dose-effect and dose-response

No dose-response data from controlled studies on man have been found in the literature. Dose-effect relations are also difficult to establish due to poor exposure data. Effects of pyridine on man are summarized in Table 1.

Some dose effect data from animal studies are shown in Tables 2 and 3.

Table 1. Summary of reported effects of pyridine on man

Exposure levels	Effect	Ref.
0.17 ppm	Odor threshold	(4)
10 ppm	Objectionable odor	(60)
700 ppm	Nose irritation threshold	(4, 65)
Occupational exposure 6-13 ppm	CNS symptoms, e.g. headache, nervousness, sleeplessness, and occasional digestive troubles	(67)
125 ppm, 4h/day for 1 to 2 weeks	Transient symptoms from the CNS and GI tract	(60)
Inhalation of unknown dose	Acute narcosis	(13)
Inhalation during 15-20 min. of unknown dose	Speech disorders 10h later	(60)
Occupational exposure to crude pyridine for six months	Disturbance of equilibrium, facial paralysis, attacks of loss of consciousness. Reversible effects.	(44)
Occupational exposure for 2 years	Wernicke's pseudo-encephalitis	(44)
1.85-2.4 ml p.o. daily for 10-30 days	Anorexia, nausea, occasional vomiting and headache, weakness, depression.	(58)
1.85-2.4 ml p.o. daily for up to two months	These were two epileptics. One died as a consequence of pyridine anti- convulsant treatment. Liver and kidney damage.	(58)
Half a cup p.o.	Died after 43h. Inflammatory lesions in the respiratory tract, the oeso- phagus and the stomach.	(31)
500mg/kg p.o.	LD <sub>Lo</sub>	(48)

Table 2. Some dose-effect data for animals exposed to pyridine

Exposure	Species	Effect	Ref.
1500 mg/kg p.o	mouse	LD50	(49)
1120 mg/kg, dermal	rabbit	Lethal dose	(7)
891 mg/kg p.o.	rat	LD50	(49)
880 mg/kg i.v.	dog	LD50	(49)
360 mg/kg i.v.	rat	LD50	(49)
330 mg/kg i.p. duration unknown	rat	Altered motor behaviour	(17)
180 mg/kg s.c., twice weekly, 1 year	rat	Maximal tolerated dose	(45)
110 mg/kg i.v.	rabbit	Narcosis	(13, 58)
100 mg/kg i.v.	rat	Alterations of bioelectrical activities	(17)
75 mg/kg i.v.	rat	Bradycardia, hypotension, changes in the ECG	(43)
50 mg/kg p.o., 90 days	rat	Some inflammatory hepatic lesions	(5)
12.5-25 mg/kg	rat	ED <sub>50</sub> , unconditioned reflex response	(66)
0.25 mg/kg, intragastric, 100 days	rat	Decrease in body weight, liver function disturbance	(66)
2.0 µg/ml in drinking water, 3 months	mouse	Brain lipid peroxidation	(57)

Table 3. Some dose-effect data for animals exposed to pyridine vapor

Exposure	Species	Effect	Ref.
23200 ppm, 1.5 h	rat	Lethal dose	(60)
4000 ppm, 4 h	rat	LD <sub>Lo</sub>	(49)
1545-3090 ppm, 40 min	rat	Changes in ammonia metabolism	(66)
309 ppm, 3h/day, 4 months	guinea pig	Decrease in body weight and body temperature, hypochromic anemia, liver damage	(66)
10 or 50 ppm, 7h/day, 5 days/week for 6 months	rat	Increased relative liver weight	(8, 28)
4.6 ppm, 60 days	rat	Reduction of phospholipids in alveolar cells and in the surface-active lining of the alveoli	(56)
0.3 ppm, 2 months	rat	Minor effects on CNS function and blood proteins	(41)

### Risk evaluation

Occupational exposure of pyridine is mainly through inhalation of its vapors. Pyridine may also enter the body through the skin. Accidental ingestion of pyridine has occurred. There are no data that indicate that pyridine accumulates in the body.

Exposure to pyridine vapors is irritating to the eyes and respiratory tract. Liquid pyridine is also irritating to the skin. Pyridine has moderate acute toxicity in mammals, including man. Long-term toxicity data are scarce. Mild symptoms of CNS injury (e.g., nausea, headache) may result from exposure to approximately 10 ppm. In rats, exposure to 10 ppm for three months caused increased liver weights. Most individuals can detect pyridine at a concentration in air of less than 1 ppm. The odor is not a reliable warning property though, since olfactory fatigue occurs quickly. High doses of pyridine cause liver and kidney damage.

The majority of short-term tests for mutagenicity and genotoxicity of pyridine are negative. The results of a chronic bioassay for carcinogenesis of pyridine in rat and mice, performed by the NTP and completed in 1985, have not yet been published. According to (24) a positive trend for mononuclear cell leukemia was found in the 2-year study, indicating that pyridine might accelerate the progression of this spontaneous (in Fischer rats) tumor.

The critical effect of exposure to pyridine is its effects on the central nervous system and the liver. The unpleasant, nauseating odor might be a problem at even lower exposures.

## Summary

Warholm M. DEC and SCG Basis for an Occupational Health Standard. Pyridine. *Arbete och Hälsa* 1991;49, pp 1-20.

Pyridine is used as a solvent and as an intermediate in the production of e.g. pharmaceutical and agricultural chemicals. Occupational exposure is mainly through inhalation of its vapors. Pyridine has an unpleasant, nauseating odor and it is irritating to the eyes, skin and respiratory tract. Mild symptoms of CNS injury (e.g., nausea, headache) may result from exposure to approximately 10 ppm. Higher doses cause liver and kidney damage. Most studies on mutagenicity and genotoxicity are negative. The critical effect of pyridine seems to be its effects on the central nervous system and the liver.

Key words: Pyridine, Occupational standards, Irritation, Odor, CNS effects, Liver effects.



## Sammanfattning

Warholm M. DEC and SCG Basis for an Occupational Health Standard. Pyridine. Arbete och Hälsa 1991:49, sid 1-20.

Pyridin används som lösningsmedel samt som intermediär vid tillverkning av bl.a. läkemedel och jordbrukskemikalier. Yrkesmässig exponering sker huvudsakligen genom inandning av pyridinångor. Pyridin har en skarp och obehaglig lukt samt är irriterande för ögon, hud och andningsvägar. Exponering för ca. 10 ppm kan påverka det centrala nervsystemet och ge upphov till huvudvärk och illamående. Högre doser orsakar skador på lever och njurar. De flesta studier över mutagenitet och genotoxicitet har varit negativa. Den kritiska effekten av pyridin tycks vara påverkan på det centrala nervsystemet och levern.

Nyckelord: Pyridin, Hygieniskt gränsvärde, Irritation, Lukt, CNS-påverkan, Leverpåverkan.

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Asbest, Evaluatie van risico op kanker bij beroepshalve blootstelling aan (aanvullend op RA 1/84)	9/89
Benzeen	5/89
Beryllium and Beryllium compounds	4/88
Blootstelling, Gezondheidskundige aspecten van het begrip en van het meten/schatten ervan	8/90
Butadiene(1,3-)	5/90
Cadmium	5/80
Caprolactam	4/84
Carbon disulphide	9/92
Carbon monoxide	7/92
Carbonylfluoride and PTFE Pyrolysis products	3/88
Carcinogene stoffen	3/80
Chloor	6/80
Chloroform	7/87
Chroom en chroomverbindingen	6/85
Cyclohexane	15/90
Cyclohexanol	3/90
Dibroomethaan	5/87
Dichloorethaan (1,1-)	8/87
Diisocyanates	3/91
Dimethyl- en diethylsulfaat	12/90
Dimethylamine	10/90
Dimethylhydrazine	2/87
Dinitro-ortho-cresol (4,6-)	4/87
Dioxaan (1,4-)	1/87
Epichloorhydrine	1/86
Ethyl acrylate	6/90
Ethyl acetate	10/91
Ethyl Methanesulphonate (EMS)	4/89
Ethyl amine	7/90
Ethylbenzene	9/91
Ethyleenoxide	6/89
Fenylhydrazine	2/87
Fluorcarbons(except FC11)	15/87
Fluorine compounds(inorganic)	1/89
Fluorine	1/89
Formaldehyde	3/87
Fosfine	1/80



Fijn hinderlijk stof; gezondheidskundige  
 aspecten van bijlage 3 bij de Nationale  
 MAC-lijst 1989

Gasoline	9/90
Hexaan (n-)	3/92
Hexachlorobenzene	11/87
Hexanone (2-)	2/88
Hydrazine	2/90
Hydrogenfluorine	2/87
Hydroxyethylhydrazine	1/89
Isopropylglycidylether	2/87
Isopropoxyethanol (2-)	1/92
Koolmonoxide (Carbon monoxide)	12/87
Kwikalkylverbindingen - Korte keten	2/79 (7/92)
Kwikverbindingen (Organische)	5/82
Lachgas (Nitrous oxide)	4/82
Lasrook (Arc welding fume....nickel)	2/85 (2/92)
Mangaan	1/93
Metallisch Kwik	1/82
Methyl acrylate	5/81
Methyleenchloride (Methylene chloride)	1/90
Methyl ethyl ketone	1/83 (8/92)
Methyl isobutyl ketone	16/90
Methyl Methanesulphonate (MMS)	4/91
Methylbromide	4/89
Monochloorethaan	13/90
Monoketones (7/8 Carbon chain Aliphatic)	2/82
n-Heptaan	14/90
Nikkel en nikkelverbindingen	1/81
Nitropropan (2-)	3/85
Nitrous oxide	1/85
Ozone	2/92
Para-Dichloorbenzeen	4/92
Pentaaan	1/88
Phthalic anhydride	2/81
Piperazine	3/89
Polyvinyl chloride (PVC) dust	7/91
Propoxyethanol (2-)	2/93
Propoxyethylacetate (2-)	12/87
Pyridine	12/87
Selenium en verbindingen	3/93
Silicon dioxide, Crystalline forms of	7/89
Stikstofdioxide	5/92
Styreen	5/85
Talc dusts	8/89
Tetrahydrofuran	6/91
Thiourea	1/91
Tolueen Diisocyaan	11/90
Tolueen	4/80
Trichloorethaan(1,1,1-)	2/91
Trichloorethyleen	3/81
Trichlorofluoromethane	3/83
Triethylamine	14/87
Trimethylamine	2/83
Vanadium metaal en anorganische verbindingen	9/87
Wood dust	10/87
Xylene	8/91
Zwaveldioxide	5/91
	4/85

## Publikaties arbeidsinspectie

### Publikatiebladen

CODE		PRIJS
P 1	Inhoud verbandtrommels Middelen voor de Eerste hulp bij ongevallen op het werk Vijfde druk 1993	f 7,50
P 41	Zittend en staand werk, ergono- mische aspecten Vierde druk 1993	f 16,-
P 77	Het tegengaan van beroepshuidaan- doeningen Tweede druk 1983	f 10,-
P 88-2	Gevaarlijke stoffen in de haven; Veilig stuwen in containers; Veilig stuwen van containers Eerste druk 1984	f 13,-
P 89	Blauwzuur; Veilige behandeling in de haven Derde druk 1983	f 10,-
P 90	Zwavelwaterstof; Veilige behandeling in de haven Tweede druk 1982	f 10,-
P 91	Zwavelkoolstof; Veilige behandeling in de haven Tweede druk 1982	f 10,-
P 92	Chloor; Veilige behandeling in de haven Derde druk 1982	f 10,-
P 93	Chloorwaterstof; Veilige behande- ling in de haven Tweede druk 1982	f 10,-
P 94	Fosgeen; Veilige behandeling in de haven Derde druk 1984	f 11,-
P 95	Allylalcohol; Veilige behandeling in de haven Derde druk 1984	f 11,-
P 96	Fluor; Veilige behandeling in de haven Tweede druk 1982	f 10,-
P 97	Fluorwaterstof; Veilige behandeling in de haven Derde druk 1984	f 11,-
P 98	Acetoncyaanhydrine; Veilige behandeling in de haven Tweede druk 1984	f 11,-
P 99	Chloorpicrine; Veilige behandeling in de haven Tweede druk 1984	f 11,-
P 100	Acrylnitril; Veilige behandeling in de haven Tweede druk 1981	f 10,-
P 101	Zwaveldioxide; Veilige behandeling in de haven Tweede druk 1982	f 10,-
P 102	Epichloorhydrine; Veilige behande- ling in de haven Tweede druk 1984	f 11,-
P 103	Allylchloride; Veilige behandeling in de haven Tweede druk 1984	f 11,-
P 104	Broom; Veilige behandeling in de haven Derde druk 1986	f 11,-
P 105	Broomwaterstof; Veilige behandeling in de haven Tweede druk 1984	f 10,-
P 107	Acetonitril; Veilige behandeling in de haven Tweede druk 1986	f 11,-
P 108	Methylbromide; Veilige behande- ling in de haven Derde druk 1984	f 11,-

## Chemie (gevaarlijke stoffen)

CODE		PRIJS
P 109	Acroleïne; Veilige behandeling in de haven Derde druk 1984	f 11,-
P 110	Loodakylverbindingen; Veilige behandeling in de haven Tweede druk 1984	f 11,-
P 111	Paration; Veilige behandeling in de haven Tweede druk 1984	f 11,-
P 112-1	Ademhalingsbeschermingsmiddelen; Overzicht en toepassing Vierde druk 1985	f 14,-
P 112-2	Ademhalingsbeschermingsmiddelen; Overzicht en beschrijving Eerste druk 1985	f 20,-
P 112-3	Keuzetabel Ademhalings- beschermingsmiddelen Tweede druk 1983	f 15,-
P 130	Laboratoria; Veiligheid bij gebruik van gevaarlijke stoffen Eerste druk 1982	f 26,-
P 130-1	Laboratoria; Veiligheid en hygiëne Algemeen Tweede druk 1982	f 16,-
P 134-1	Zweminrichtingen; Wettelijke bepalingen Derde druk 1982	f 11,-
P 134-2	Zweminrichtingen; De opslag en het gebruik van natrium hypochlo- riet (=—chloorbleekloog) Vierde druk 1988	f 11,-
P 134-3	Zweminrichtingen; De opslag en het gebruik van zoutzuur Vierde druk 1988	f 11,-
P 134-4	Zweminrichtingen; De opslag en het gebruik van zwavelzuur Vierde druk 1988	f 11,-
P 134-5	Zweminrichtingen; De opslag en het gebruik van kooldioxide Eerste druk 1980	f 12,-
P 139	Verfverwerking Tweede druk 1986	f 13,-
P 145	Nationale MAC-lijst 1992 Achtste druk 1992	f 16,-
PMAC	Zakboek MAC-waarden 1989	f 9,-
P 167	Chemisch reinigen van textiel Eerste druk 1987	f 13,-
P 171-1	Vaklokalen en theorievaklokalen; Scheikunde Eerste druk 1988	f 14,-
P 172-1	Arbeidsveiligheidsrapport; Leidraad aanwijzing AVR-plichtige installaties Eerste druk 1988	f 14,-
P 172-1E	Occupational Safety Report; Designatory guidelines for AVR- mandatory installations First edition 1988	f 14,-
P 172-2	Arbeidsveiligheidsrapport; Leidraad voor het samenstellen Eerste druk 1989	f 15,-
P 172-2E	Occupational Safety Report; Guide- line for compilation Eerste druk 1990	f 15,-
P 182	Gevarenzone-indeling met betrek- king tot gasontploffingsgevaar Eerste druk 1992	f 27,-
P 184	Werken met beeldschermen Eerste druk 1992 ISBN 9039903735	f 20,-

## Concept-publikatiebladen

CODE		PRIJS
CP 1	Het werken met recombinant DNA in CI- en CII-laboratoria Eerste druk 1980	f 19,-
CP 8	Veilig werken met PCB's en apparaten die PCB's bevatten Eerste druk 1985	f 20,-
CP 19	Tapruimten en taplokalen voor gevaarlijke stoffen Eerste druk 1990	f 21,-
CP 22	Tankauto's; Laden en lossen van gevaarlijke stoffen Eerste druk 1990	f 15,-
CP 35	Productiekennisgeving in het kader van het kennisgevingsstelsel Wet Milieugevaarlijke Stoffen Eerste druk 1992	f 17,50
CP 36	Ergonomische richtlijnen voor informatieoverdracht bij procesbesturing Eerste druk 1992	f 22,50

## Voorlichtingsbladen

CODE		PRIJS
V 1-E	The impact op modifications in the process-industry on safety Eerste druk 1981	f 11,-
V 5	Technische inspectie van installaties in de procesindustrie; Enkele principes en achtergronden Eerste druk 1982	f 21,-
V 5-E	Technical inspection of installations in the process-industry; Basic principles and background Eerste druk 1983	f 21,-
V 6	Instrumentele beveiligings- en gevaardetectiesystemen in de procesindustrie; Enkele principes en grondslagen Eerste druk 1984	f 29,-
V 7	Procesveiligheidsanalyse; Aanzet tot het opsporen van inherente procesgevaaren Eerste druk 1984	f 31,-
V 7-E	Process safety analysis; Incentive for the identification of inherent process hazards Eerste druk 1985	f 30,-
V 9	Beroepshuandaandoeningen; Handleiding onderzoek Eerste druk 1985	f 32,-
V 11	Checklist procesinstallaties; Aandachtspunten voor een veilig ontwerp Tweede druk 1986	f 31,-
V 11-E	Checklist processing plants Tweede druk 1989	f 26,-
V 14	Explosiebestendige controlegebouwen in de procesindustrie Eerste druk 1977	f 11,-
V 18	Procedures in de procesindustrie; Voorbeelden en voorstellen met betrekking tot het ontwikkelen, invoeren en beheren van procedures in de procesindustrie Eerste druk 1989	f 25,-

CODE PRIJS

V 18-E	Procedures in the process industry; Examples and proposals concerning development, introduction and control of procedures in the process-industry Eerste druk 1989	f 21,-
V 23	Gevaaren van statische elektriciteit in de procesindustrie Tweede druk 1991 ISBN 9053071903	f 36,-

## Concept-voorlichtingsbladen

CODE		PRIJS
CV 2	Hoe te handelen bij blootstelling aan longbeschadigende gassen of dampen Eerste druk 1981	f 21,-
CV 4	Acrylnitril; Technisch-arbeids-hygiënische voorzieningen Eerste druk 1983	f 26,-
CV 12	Ademhalingsbescherming; Fysiologische en gezondheidsaspecten Eerste druk 1989	f 24,-
CV 14	Veiligheid van gebouwen in de procesindustrie; Aandachtspunten bij ontwerp, constructie en gebruik Eerste druk 1989 ISBN 9053070028	f 39,-
CV 21	Procesbeheersingssystemen aandachtspunten voor een veilige en correcte toepassing Eerste druk 1991 ISBN 905307192X	f 34,-
CV 28	Afgasbehandelingssystemen Eerste druk 1992 ISBN 9053072993	f 25,-

## Studies

CODE		PRIJS
S 20	Acute intoxicaties in de werksituatie April 1986	f 20,-
S 20-1	Acute intoxicaties in de werksituatie Februari 1989	f 20,-
S 20-2	Acute intoxicaties in de arbeids-situaties in 1989 Maart 1991	f 19,-
S 20-3	Acute intoxicaties in de arbeids-situaties in 1990 September 1991	f 20,-
S 28	Blaastumoren als beroepsziekten ten gevolge van blootstelling aan chemische stoffen Oktober 1986	f 34,-
S 29-1	Chronische effecten tengevolge van blootstelling aan organische oplosmiddelen; Samenstelling, voorkomen, blootstellingsniveaus en effecten December 1986	f 26,50
S 29-2	Chronische effecten tengevolge van blootstelling aan organische oplosmiddelen; Effecten van blootstelling aan organische oplosmiddelen op het centrale zenuwstelsel November 1986	f 21,50
S 30	Protocollen voor de bedrijfsgezondheidszorg; Verzamelband	f 12,-
S 30-1	Algemeen April 1987	f 19,-
S 30-2	Benzeen April 1987	f 9,-
S 30-3	Styreen April 1987	f 9,-

CODE		PRIJS	CODE		PRIJS
S 30-4	Xylenen April 1987	f 9,-	CPR 3-E	Storage of organic peroxides Eerste druk 1982	f 30,-
S 30-5	Trichlooretheen April 1987	f 9,-	CPR 4	Experimenten met chloor (rapporten) Eerste druk 1979	f 46,-
S 30-6	Tetrachlooretheen April 1987	f 9,-	CPR 5	Vloebare zuurstof; Opslag van 0,45 -100 m3 Eerste druk 1983	f 21,-
S 30-7	Zwavelkoolstof April 1987	f 9,-	CPR 6	Vloebare zwaveldioxide Eerste druk 1983	f 32,50
S 30-8	Lood en zijn ionverbindingen Juni 1988	f 9,-	CPR 7	De bewaring van springstoffen en ontstekingsmiddelen Eerste druk 1983	f 14,-
S 30-9	Chloroform Juni 1989	f 9,-	CPR 8-1	Supplement Autogas (LPG) Supplement 1988	f 22,50
S 30-10	Molybdeen en anorganische molybdeenverbindingen Juni 1989	f 9,-	CPR 8-2	LPG-Tankwagens Eerste druk 1985	f 27,50
S 30-11	Metallisch kwik (A); Kwikzouten (B); Organische kwikverbindingen Juni 1989	f 9,-	CPR 8-3	Distributiedepots voor LPG (Butaan, Propaan en hun mengsels) Eerste druk 1991 ISBN 9053071652	f 42,50
S 30-12	Mangaan en anorganische man- gaanverbindingen Juni 1989	f 9,-	CPR 9-1	Vloebare aardolieproducten; Ondergrondse opslag in stalen tanks en afleverinstallaties voor motor-brandstof Concept vijfde druk 1991	f 26,-
S 30-13	Antimoon en antimoon- verbindingen Juni 1989	f 9,-	CPR 9-2	Vloebare aardolieproducten; Bovengrondse opslag kleine installaties Eerste druk 1985	f 29,-
S 30-14	Tetraethyllood; Tetramethyllood Juni 1989	f 9,-	CPR 9-3	Vloebare aardolieproducten; Bovengrondse opslag grote installaties Eerste druk 1984	f 36,-
S 30-15	Methanol Juli 1990	f 9,-	CPR 9-5	Vloebare aardolieproducten; Ondergrondse opslag van vloebare producten in kunst- stof tanks Eerste druk 1992 ISBN 9039903727	f 27,50
S 30-16	Fenol Juli 1990	f 9,-	CPR 10	Chloor; Opslag en gebruik Tweede druk 1983	f 36,-
S 30-17	n-Hexaan Juli 1990	f 9,-	CPR 11-1	Propaan; Het gebruik van propaan op bouwterreinen Eerste druk 1984	f 35,-
S 30-18	Methyl-n-butylketon (2-hexanon) Juli 1990	f 9,-	CPR 11-2	Propaan (5 m3); De opslag van propaan en butaan in stationaire bovengrondse reservoirs met een inhoud groter dan 0,15 m3 en ten hoogste 5 m3 Eerste druk 1986	f 34,-
S 30-19	Isopropylbenzeen (cumeen) Juli 1990	f 9,-	CPR 11-3	Propaan; Opslag van propaan en butaan in stationaire boven- grondse- en terpreservoirs met een inhoud groter dan 5 m3 en ten hoogste 150 m3 Eerste druk 1990 ISBN 9053070591	f 36,-
S 47	Werkterreinanalyse van trichloor- monofluormethaan (freon-11) December 1988	f 21,50	CPR 11-4	Propaan; Toepassing van propaan in wegenbouwmachines en onkruidbestrijdingsmachines Eerste druk 1990 ISBN 9053070605	f 19,-
S 48	Werkterrein-analyse van hexa- chloobenzeen December 1988	f 12,50	CPR 11-5	Propaan vulstations van butaan- en propaanflessen Concept eerste druk 1991	f 41,-
S 49	Werkterreinanalyse van 1,3-buta- dien December 1988	f 19,-	CPR 12	Methoden voor het bepalen en verwerken van kansen Eerste druk 1985	f 127,50
S 50	Beroepsmatige blootstelling aan organische stof en de daarmee samenhangende risico's voor de gezondheid December 1988	f 47,50	CPR 12-E	Methods for determining and processing probabilities Eerste druk 1988	f 127,50
S 62	Werkterreinanalyse van anorga- nische oplosbare fluoriden April 1989	f 26,50	CPR 13	Ammoniak; Vervoer, opslag en toepassingen Tweede druk 1988	f 46,-
S 63	Werkterreinanalyse van 1,2-di- chloor-ethaan April 1989	f 20,-	CPR 14	Methoden voor het berekenen van fysische effecten Eerste druk 1988	f 159,-
S 64	Werkterreinanalyse van aniline Tweede druk 1991 ISBN 9053071415	f 21,50	CPR 14-E	Methods for the calculation of physical effects Tweede druk 1991	f 156,-
S 65	Werkterreinanalyse van cyclohexanol April 1989	f 17,50			
S 116	Arbeidsomstandigheden in de chemische industrie April 1991 ISBN 9053071628	f 35,-			

### Commissie preventie van rampen door gevaarlijke stoffen

CODE		PRIJS
CPR 2	Model voor risico-evaluatie van opslag van gevaarlijke stoffen, vloeistoffen en gassen Eerste druk 1982	f 19,-
CPR 3	Organische peroxiden; Opslag Eerste druk 1982	f 29,-

CODE		PRIJS
CPR 15-1	Opslag gevaarlijke stoffen in emballage; Opslag van vloeistoffen en vaste stoffen (0 tot 10 ton) Tweede druk 1990 ISBN 9053070338	f 20,-
CPR15-1E	Storage of Packaged Hazardous Materials; Storage of liquids and solids (0-10 tons) Eerste druk 1992	f 19,-
CPR 15-2	Opslag gevaarlijke stoffen, chemische afvalstoffen en bestrijdingsmiddelen in emballage; opslag van grote hoeveelheden Eerste druk 1991 ISBN 9053072128	f 22,50
CPR 15-3	Opslag van bestrijdingsmiddelen in emballage; Opslag van bestrijdingsmiddelen in distributiebedrijven en aanverwante bedrijven (vanaf 400 kg) Eerste druk 1990 ISBN 9053071024	f 21,-
CPR15-3E	Storage of Packaged pesticides Storage of pesticides in distribution and related enterprises (in excess of 400 kg) Eerste druk 1992	f 21,-
CPR 16	Methoden voor het bepalen van mogelijke schade aan mensen en goederen door het vrijkomen van gevaarlijke stoffen Eerste druk 1990	f 159,-
CPR 16E	Methods for the determination of possible damage to people and objects resulting from releases of hazardous materials Eerste druk 1992	f 159,-

## R-bladen

CODE		PRIJS
R 1	Voorlopige richtlijnen voor de beveiliging van stookinstallaties met een maximum belasting groter dan 600 kW in de procesindustrie en die gestookt worden met gasvormige of vloeibare brandstoffen Eerste druk 1978	f 29,-
R 1-E	Provisional guidelines for the safeguarding of fuel-burning installations with a maximum try and fired gaseous or liquid fuels Eerste druk 1979	f 36,-
R 2-E	Guide for the classification of hazardous areas in zones in relation to gasexplosion hazards and to the installation and selection of electrical apparatus Eerste druk 1980	f 31,-
R 3-E	Hazard and operability study; Why? When? How? Eerste druk 1979 (Nederlandse uitgave zie V2)	f 24,-

## Overige publikaties

CODE		PRIJS
OP 1	Experimenten met acrylnitril Eerste druk 1971	f 19,-
OP 1-E	Experiments with Acrylonitrile Eerste druk 1972	f 19,-
OP 2	Experimenten met acrylnitril; Blusproeven (Nederlands en Engels) Eerste druk 1972	f 20,-
OP 3	Experimenten met chloor Eerste druk 1975	f 16,-

CODE		PRIJS
OP 3-E	Experiments with chlorine Eerste druk 1975	f 16,-
OP 9	Leidraad voor oliepijpleidingen Eerste druk 1973	f 21,-
OP 12	De opslag en het vervoer van acrylnitril Eerste druk 1970	f 22,50
OP 14	De opslag en het gebruik van fosgeen Eerste druk 1977	f 31,-
OP 17	Voorlopige richtlijn voor de beveiliging van met olie/aardgas gestookte éénbranderinstallaties met een maximum belasting groter dan 600 kW Eerste druk 1975	f 13,-

## Gezondheidskundige adviezen van de werkgroep van deskundigen ter vaststelling van MAC-waarden

CODE		PRIJS
RA 2/79	Grenswaarde Koolmonoxyde	f 29,-
RA 1/80	Grenswaarde Fosfine	f 15,-
RA 2/80	Grenswaarde anorganisch lood	f 22,50
RA 3/80	Grenswaarde Carcinogene stoffen	f 20,-
RA 4/80	Grenswaarde Tolueen Diisocyaanaat	f 9,-
RA 5/80	Grenswaarde Cadmium	f 20,-
RA 6/80	Grenswaarde Chloor	f 16,-
RA 1/81	Grenswaarde n-Heptaan	f 14,-
RA 2/81	Grenswaarde Pentaan	f 11,-
RA 3/81	Grenswaarde 1,1,1,-Trichloorethaan	f 22,50
RA 5/81	Grenswaarde metallisch kwik	f 16,-
RA 1/82	Grenswaarde Mangaan	f 21,-
RA 2/82	Grenswaarde Monochloorethaan	f 14,-
RA 3/82	Grenswaarde anorganische kwikzouten	f 19,-
RA 4/82	Grenswaarde organische kwikverbindingen (uitsluitend phenylkwik en alkoxyalkylverbindingen)	f 16,-
RA 5/82	Grenswaarde kwikalkylverbindingen korte keten (uitsluitend methylkwik en ethylkwik)	f 22,50
RA 1/83	Grenswaarde Methyleenchloride	f 21,-
RA 2/83	Grenswaarde Triethylamine	f 20,-
RA 3/83	Grenswaarde Trichloorethyleen	f 22,50
RA 1/84	Asbest	f 35,-
RA 2/84	Grenswaarde anorganische Arseenverbindingen (exclusief Arseenwaterstof)	f 25,-
RA 4/84	Grenswaarde Caprolactam	f 21,-
RA 1/85	Grenswaarde 2-Nitropropan	f 15,-
RA 2/85	Grenswaarde Lachgas	f 26,-
RA 3/85	Grenswaarde Nikkel en nikkelverbindingen	f 26,-
RA 4/85	Grenswaarde Zwaveldioxide	f 21,-
RA 5/85	Grenswaarde Stikstofdioxide	f 19,-
RA 6/85	Grenswaarde Chroom en chroomverbindingen	f 25,-
RA 1/86	Grenswaarde Epichloorhydrine	f 24,-
RA 1/87	Grenswaarde 1,4-Dioxaan	f 16,-
RA 2/87	Grenswaarde Hydrazine, dimethylhydrazine, hydroxyethylhydrazine en fenylhydrazine	f 26,-
RA 3/87	Grenswaarde Formaldehyde (engelse uitgave)	f 27,50
RA 4/87	Grenswaarde 4,6-Dinitro-orthoecresol	f 16,-
RA 5/87	Grenswaarde Dibroomethaan	f 16,-
RA 6/87	Grenswaarde Aflatoxine B1, B2, G1 en G2	f 20,-

CODE		PRIJS
RA 7/87	Grenswaarde Chloroform	f 22,50
RA 8/87	Grenswaarde 1,1-Dichloorethaan	f 11,-
RA 9/87	Grenswaarde Trimethylamine	f 16,-
RA 10/87	Grenswaarde Vanadium metaal en anorganische verbindingen	f 20,-
RA 11/87	Grenswaarde n-Hexaan	f 26,-
RA 12/87	Grenswaarde 2-Propoxyethanol, 2-Propoxyethyl acetate, 2-Isopropoxyethanol	f 11,-
RA 13/87	Grenswaarde Acrilaten	f 16,-
RA 14/87	Grenswaarde Trichlorofluoromethane	f 20,-
	(engelse uitgave)	
RA 15/87	Grenswaarde Fluorcarbons (except FC11)	f 26,-
	(engelse uitgave)	
RA 1/88	Grenswaarde Para-Dichloorbenzeen	f 19,-
RA 2/88	Grenswaarde Hexachloorbenzeen	f 30,-
	(engelse uitgave)	
RA 3/88	Grenswaarde Carbonylfluoride and PTFE Pyrolysis products	f 14,-
	(engelse uitgave)	
RA 4/88	Grenswaarde Beryllium and Beryllium compounds	f 27,50
	(engelse uitgave)	
RA 1/89	Grenswaarde Fluorine, Hydrogenfluoride and Inorganic Fluoride Compounds	f 27,50
	(engelse uitgave)	
RA 2/89	Grenswaarde Aniline	f 21,-
	(engelse uitgave)	
RA 3/89	Grenswaarde Phthalic anhydride	f 15,-
	(engelse uitgave)	
RA 4/89	Grenswaarde Ethyl Methanesulphonate (EMS) Methyl Methanesulphonate (MMS)	f 27,50
	(engelse uitgave) ISBN 9053070354	
RA 5/89	Grenswaarde Benzeen	f 13,-
	ISBN 9053070362	
RA 6/89	Grenswaarde Ethyleenoxide	f 16,-
	ISBN 9053070370	
RA 7/89	Grenswaarde Selenium en verbindingen	f 22,50
	ISBN 9053070389	
RA 8/89	Grenswaarde Styreen	f 21,-
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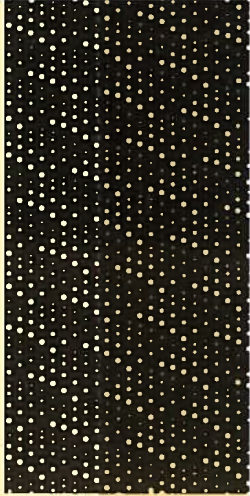
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